

## ORIGINAL ARTICLE OPEN ACCESS

# A Comparative Study of *Litchi-chinensis* Biodiesel–Diesel Blends With Alcoholic and Nanoparticle Additives on Engine Performance, Emissions, and Coefficient of Friction

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## ABSTRACT

Environmental hazards and the growing demand for fossil fuels are two major current concerns. The current global situation requires extensive research on renewable fuels. In this experimental study, we investigated different blends of biodiesel derived from litchi seed oil, which is inedible, for engine performance, emissions, and tribological characteristics. *Litchi chinensis* biodiesel was synthesized from litchi seed oil through a microwave-assisted transesterification. First, a 30% biodiesel–diesel blend (B30) was prepared and mixed with ethyl ethanoate (EE) in the proportions of 10% (B30EE10) and 15% (B30EE15) by volume. Additionally, 100 ppm of aluminum oxide (abbreviated as Al) nanoparticles were added in all blends (B30Al, B30EE10Al, and B30EE15Al). All six blends were investigated for engine performance and emissions on a diesel engine. The coefficient of friction (COF) was measured using a four-ball tribo-tester for a 5% by volume fuel blend in SAE-40 lubricant (ASTM D4172 standard). Engine performance analyses indicated that, compared to diesel, the B30EE10Al gives most promising results at an engine speed of 1500 min<sup>−1</sup> in terms of engine performance and emissions with 4.97% higher brake thermal efficiency and by reducing brake-specific fuel consumption, and CO, HC, and NO<sub>x</sub> up to 15.27%, 30.24%, and 3.92%, respectively. Four-ball tribo-tester results reveal that 5% B30Al in SAE-40 oil has the lowest COF of 0.066, which is 34% less than that of diesel. Overall, the addition of aluminum nanoparticles and EE10 in *Litchi chinensis* biodiesel blend has improved the fuel quality in terms of performance and emission properties.

## 1 | Introduction

The increasing fossil fuel consumption to meet the energy requirements of transportation vehicles, which are heavily dependent on petrol and diesel engines, is a major global issue [1]. Diesel engines, being more energy-efficient and cost-effective, are utilized more often compared to their petrol counterparts [2]. Because of the

rapidly depleting fossil fuels and the severe consequences of greenhouse gases (GHGs), researchers are looking for innovative energy solutions and fuel sources that can be helpful in the face of this global challenge [3, 4].

Biodiesel is an excellent alternative fuel for diesel engines because of its superior physicochemical properties [5]. This

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concept has led to the use of many novel resources for biodiesel production, like *Prosopis juliflora* [6], *Pontederia crassipes*, and *Trichosanthes cucumerina* [7], etc. However, the controversy over fuel versus food and the expensive cost of feedstock for biodiesel synthesis make the nonedible or waste material feedstock more desirable as an alternative energy resource [8]. A recent review found that nonedible oils can be used as a biodiesel feedstock, improve storability, engine performance, and physical attributes of diesel blends [9].

Numerous researchers have looked at various renewable bio-fuels, such as alcohols, vegetable oils, and biodiesel [10–13]. In this proposed study, *Litchi chinensis* seed biodiesel is being used to prepare a fuel blend with 30% biodiesel in 70% diesel by volume (B30). The litchi seed contains methylene cyclopropyl-alanine, which is toxic in nature and thus cannot be used as a food source [14]. The oil extracted from the starch of litchi seeds can be processed and utilized as biofuel, which not only reduces the environmental damage due to tailpipe emissions but is also workable for organic waste management [15, 16]. Fatty acid compositions have a significant impact on the cold flow characteristics and oxidation stability of biodiesel [17]. Properties of litchi biodiesel reveal high unsaturated fatty acids and good cold flow characteristics [18].

The commercialization of biodiesel has met many compositional and functional obstacles despite its superior physicochemical qualities [5, 19, 20]. The commercial feasibility of biofuels is also challenged due to poor cold flow characteristics [21]. The method of biodiesel production also affects the quality of fuel. A comparison of soybean biodiesel production reveals that fuel obtained from transesterification has produced better results in terms of engine performance and emission characteristics than that of the pyrolysis method [22]. However, some fuels have been found to have better properties than others. Castor oil has a workable cloud point and viscosity because it contains fatty acid ethyl ester (FAEE) [23]. Because of lower pour and cloud point values, castor-based biodiesel performs well in cold areas and has good lubricity [24]. Litchi biodiesel is expected to have better cold flow properties due to its high unsaturated fatty acids [18].

By using innovative techniques, such as adding oxygenated alcohols and nanoparticle additives, these disadvantages can be overcome and potentially improve the performance of diesel engines running on biofuels [25, 26]. The use of additive nanoparticles and oxygenated additives (mainly alcohols) can complement both the good and poor qualities of biodiesel. Table 1 summarizes the notable discoveries from similar studies on the effects of biodiesel–diesel blends, nanoparticles, and alcoholic additives on diesel engines.

Fuel mix properties may differ in combination with other fuels; a 20% palm biodiesel–diesel blend mixed with *n*-butanol and diethyl ether gives higher brake thermal efficiency (BTE) than the same additive concentrations in jatropha biofuel. Similarly, ethanol demonstrated a visible reduction in emission compared to *n*-butanol and diethyl ether [35]. Combustion of palm biodiesel with diesel shows a drop in brake-specific fuel consumption (BSFC), which further reduces with oxyhydrogen (HHO) gas inclusion, indicating that the addition of both oxygen and hydrogen improves the combustion efficiency of fuel

blends [36]. Ten percent by volume of dimethyl carbonate (DMC) in a 40% sapota seed biodiesel–diesel blend resulted in lower smoke, hydrocarbon (HC), and carbon monoxide (CO) emissions when studied under 10% exhaust gas recirculation (EGR) [37]. Di-tetra-butyl-peroxide (DTBP) improves the BTE of a 50%–50% papaya seed biodiesel–diesel blend with reduced smoke, CO, and HC in exhaust gases, particularly owing to its high oxygen content [38].

The alcohols have high latent heat of evaporation; thus, their addition to the fuel blends reduces the NO<sub>x</sub> formation [39]. In comparison, the oxygenated additives are more effective in soot reduction than esters (biodiesels) having the same mass fraction of oxygen [39–41]. Higher level alcohols show more overall improvement in engine combustion, performance, and emission characteristics compared to lower alcohols (butanol) in a mix of 30% oleander methyl ester and carbon nanotubes (CNTs) [42].

Ahmad et al. [43] worked on oxygenated additives in second-generation biodiesels and concluded that the BSFC mostly increased with the inclusion of oxygenated agents. In another study, 12% diethyl ether in a 20% tamarind-seed biodiesel–diesel blend gives a significant drop in harmful emissions [44]. Similarly, 5%–15% 2-ethoxy ethyl acetate blends in diesel enhanced engine performance and implicitly lowered the tailpipe emissions [45].

Ethyl ethanoate (EE) was used in this study as an oxygenated additive. In literature, 5%–10% of EE showed a considerable drop in emissions along with a minimal rise in fuel consumption of biodiesel–diesel blend [29]. EE is a renewable fuel additive, and its high oxygen level gives the NO<sub>x</sub>–PM trade-off as well as NO<sub>x</sub>–BSFC trade-off properties without any thermal efficiency losses in a diesel engine [29, 46]. Aguado-Deblas et al. [47] studied the effects of EE in sunflower and castor oil biofuel blends and observed higher fuel consumption due to lower calorific value (CV), reduction of soot owing to low cetane number, better combustion at low temperatures, and lower emission. Different proportions (5.65%, 10.65%, and 15%) of EE in neat diesel give lower values of heat release rate and in-cylinder peak pressure and increased CO and HC. At 15% EE, engine performance was stable with less NO<sub>x</sub> emission and 15.35% less BSFC [30]. It is also observed, from the previous studies, that the use of EE as an additive in biofuels, up to 15% by volume, does not cause any issues related to engine operation. Furthermore, its good miscibility makes it desirable to use in biodiesel–diesel blends and improves the quality of the fuel mix [29]. EE also shows better cold flow properties when added with castor oil and waste cooking oil [47].

Despite significant NO<sub>x</sub> reduction qualities and high oxygen content of oxygenated additives, their low cetane number and poor lubricating qualities make them difficult to employ as a fuel enhancer in diesel engines [5, 48]. On the other hand, nanoparticles show superior results in terms of brake power (BP), BSFC, and BTE of the engine and in lowering NO<sub>x</sub> and HC [28]. Researchers have been testing the effectiveness of using biodiesel–diesel blends in place of neat diesel by employing a variety of additives in different combinations to enhance the physiochemical properties of these renewable fuels [49].

**TABLE 1** | Key findings from several studies involving the use of biodiesel–diesel blends, nanoparticles, and alcoholic additives in diesel engines and analysis of their performance, emission, and tribological characteristics. Upward arrows show an increase, and downward arrows represent a decrease.

| Findings              |                                                                                                                                                                                                                                                                           |                                                   |                                                                                                    |                         |
|-----------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|---------------------------------------------------|----------------------------------------------------------------------------------------------------|-------------------------|
| Author                | Sample blends                                                                                                                                                                                                                                                             | Performance parameters                            | Emission characteristics                                                                           | Coefficient of friction |
| Mujtaba et al. [27]   | B10 (diesel)<br>B30<br>B30CNT<br>B30TiO <sub>2</sub><br>B30DMC<br>B30DEE                                                                                                                                                                                                  | BSFC ↑ ↓<br>BTE ↑                                 | HC ↓ (except for B30 + CNT)<br>CO ↓<br><br>NO <sub>x</sub> ↑ (except for B30 + CNT)                | —                       |
| Ağbulut et al. [28]   | D100 (diesel)<br><br>B10 (90% diesel–10% of waste cooking oil biodiesel)<br>B10TiO <sub>2</sub> (B10 with 100 ppm of titanium oxide)<br>B10Al <sub>2</sub> O <sub>3</sub> (B10 with 100 ppm of aluminum oxide)<br>B10SiO <sub>2</sub> (B10 with 100 ppm of silicon oxide) | BSFC ↓ (except for B20)<br>BTE ↑ (except for B20) | CO ↓<br>NO <sub>x</sub> ↓<br>HC ↓                                                                  | —                       |
| Çakmak [29]           | D80B20 (0% ethyl ethanoate, 20% of an EN 14214 standard biodiesel, and 80% diesel)<br>D80B15EE5<br>D80B10EE10                                                                                                                                                             | —                                                 | CO ↓<br>NO <sub>x</sub> ↓                                                                          | —                       |
| Musyoka et al. [30]   | D100 (100% diesel, 0% <i>n</i> -butanol, and 0% ethyl ethanoate)<br>Bu10.65<br>D85Bu15<br>D80Bu20<br>D94.35EE5.65<br>D89.35EE10.65<br>D85EE15                                                                                                                             | BSFC ↓ ↑                                          | NO <sub>x</sub> ↓<br>CO ↑<br>HC ↑                                                                  | —                       |
| Soudagar et al. [31]  | D80HOME20 (20% honge oil methyl biodiesel, 80% diesel, and 0ppm Al <sub>2</sub> O <sub>3</sub> )<br>D80HOME20Al20<br>D80HOME20Al40<br>D80HOME20Al60                                                                                                                       | BTE ↓<br>BSFC ↑                                   | HC ↑<br>NO <sub>x</sub> ↑<br>CO ↑                                                                  | —                       |
| Fayad and Dhahad [32] | Diesel<br>B20 (0 mg/L Al <sub>2</sub> O <sub>3</sub> , 20% butanol, and 80% diesel)<br>B20Al30<br>B20Al50<br>B20Al100                                                                                                                                                     | BSFC ↓ (except for B20)<br>BTE ↑                  | NO <sub>x</sub> ↓ (except for B20, i.e., NO <sub>x</sub> ↓ ↑)<br>CO ↓<br>HC ↓<br>CO <sub>2</sub> ↑ | —                       |
| Mujtaba et al. [33]   | B10 (diesel)<br>B30                                                                                                                                                                                                                                                       | BSFC ↓ ↑<br>BTE ↑                                 | HC ↓<br>CO ↓                                                                                       | COF ↓                   |

(Continues)

TABLE 1 | (Continued)

| Author              | Sample blends                                          | Findings                                                                 |                          |                         |
|---------------------|--------------------------------------------------------|--------------------------------------------------------------------------|--------------------------|-------------------------|
|                     |                                                        | Performance parameters                                                   | Emission characteristics | Coefficient of friction |
| Mujtaba et al. [34] | B30TiO <sub>2</sub><br>B30DMC                          | Tribological samples: 5% of each fuel blend in 95% SAE 40 base lubricant | NO <sub>x</sub> ↑        | COF ↓                   |
|                     | SAE 40 lubricant                                       |                                                                          |                          |                         |
|                     | 5% B10 (diesel) + lubricant                            |                                                                          |                          |                         |
|                     | 5% B30 (30% palm-sesame [50–50] biodiesel) + lubricant |                                                                          |                          |                         |
|                     | 5% (B30DMC20) + lubricant                              |                                                                          |                          |                         |
|                     | 5% (B30Eth10) + lubricant                              |                                                                          |                          |                         |
|                     | 5% (B30CNT100) + lubricant                             | —                                                                        | —                        | —                       |
|                     | 5% (B30TiO <sub>2</sub> 100) + lubricant               |                                                                          |                          |                         |

Nanoparticle additives, usually metal-based oxides or carbon-containing compounds, are also desirable in fuel quality improvement in biodiesel [26]. Previous research has shown that carbon-based nanoparticles like CNTs and GO (graphene oxide), as well as metallic nanoparticles like CuO, ZnO, MnO, TiO<sub>2</sub>, CeO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, and Fe<sub>2</sub>O<sub>3</sub>, have shown remarkable outcomes. By achieving high BTE and high BP and lowering carbon emissions and NO<sub>x</sub>, the mentioned nanoparticle additives seem to improve engine performance [26, 50–52].

Metal-oxide-based nanoparticle additives enhance the heating value of biodiesel and improve viscosity and cetane number. A 10% waste cooking oil biodiesel blend (B10), evaluated against the inclusion of 100 ppm nanoparticles of Al<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub>, and SiO<sub>2</sub> in B10 biofuel, the maximum decrease in CO was observed with Al<sub>2</sub>O<sub>3</sub> addition due to its higher oxygen content [28]. Al<sub>2</sub>O<sub>3</sub> also improved the combustion process by enhancing in-cylinder pressure and heat release rate when studied in Juliflora oil biodiesel [53]. Two different biofuel blends, 20% honge oil biodiesel–diesel with aluminum and 20% dairy scum biodiesel–diesel blend mixed with graphene oxide nanoparticles, were investigated by Soudagar et al. [31, 54]. Their findings revealed that 40 ppm of both aluminum and graphene nanoparticles in two biodiesels reduced CO by 48.43% and 38.66%, respectively; 50 ppm of Al<sub>2</sub>O<sub>3</sub> in 20% mahua biofuel enhances the fuel atomization and thermal efficiency and reduces the emissions by 15%–30% for NO<sub>x</sub>, HC, smoke, and CO [55, 56].

The addition of 100 ppm Al<sub>2</sub>O<sub>3</sub> to a 20% butanol blend in 80% neat diesel (B20 butanol–diesel blend) improves emissions and combustion [32]. Al<sub>2</sub>O<sub>3</sub> and CeO<sub>2</sub>, 100 ppm of each when analyzed in a B20 mixture of mahua biodiesel give similar results with 100 ppm of Al<sub>2</sub>O<sub>3</sub> in a B20 butanol–diesel blend [32, 57]. On the other hand, graphical results of Al<sub>2</sub>O<sub>3</sub> and CeO<sub>2</sub>, 150 ppm each in a B20 mixture of mahua biodiesel, show increased NO<sub>x</sub>, specifically at higher loads, and an irregular trend of CO and HC emissions with increasing load, potentially caused by combustion instability [57]. Aalam et al. [58] performed an experimental study using a 25% Ziziphus jujube methyl ester (ZJME25) blend and got similar results as in Hameed and Muralidharan's work [57]. The addition of Al<sub>2</sub>O<sub>3</sub> gives further rise to the NO<sub>x</sub> emission in the ZJME25 blend. To sum it all up, the use of different fuel additives along with biodiesel blends is becoming popular in the engine research fraternity [59, 60].

Tribological properties of fuel are also important for the long-lasting operation of engines. The tribological studies are mainly concerned with minimizing coefficient of friction (COF) between mechanical surfaces [61]. Lubricant oils are used to improve tribological properties and are continuously worked on to explore enhanced characteristics to minimize wear by the inclusion of solid nanoparticles and liquid additives. Nanoparticles reduce the degradation of lubricating oil by forming a layer between metallic surfaces [34]. Study of the tribological aspects of biodiesel fuel is vital, as fuel blends can penetrate through the ring–piston–cylinder assembly and thus affect the lubricity of engines by polluting the engine lubricating oil. Literature shows that it can make up to 5% of total lubricant volume [33, 62].

According to Kumar et al. [63], castor biodiesel has better physicochemical characteristics and reduced COF when compared to synthetic 20W-40 base oil. In a study related to novel *Jasminum officinale* biodiesel blends, the lubricity of the tested fuels rises with the addition of biodiesel concentration at all the tested conditions [64]. Other studies have shown improved tribological behavior of the biofuels with MgO, graphene, and Al<sub>2</sub>O<sub>3</sub> nanoparticles [65]; 1 wt.% graphene nanoparticles in neem oil, when tested under ASTM standards and four-ball tribo-tester, a reduced COF with smoother wear scar diameter (WSD) was observed [66]. A study by Ghalme et al. [67] shows that the optimal value of 0.5 wt.% of Al<sub>2</sub>O<sub>3</sub> nanoparticles improved WSD and COF at all loading conditions. Independent of its concentrations in biodiesel–diesel blends, Al<sub>2</sub>O<sub>3</sub> had the highest antiwear performance among the nanofuels studied in filtered tire pyrolysis oil biodiesel [68]. Overall, Al<sub>2</sub>O<sub>3</sub> nanoadditives reveal the best traits even in terms of price, load, speed, and low concentration usage [65]. The performance improvement is due to the self-protective film formed by Al<sub>2</sub>O<sub>3</sub> nanoparticles on the frictional surfaces, which occurs in response to its ball bearing effect [67]. Mujtaba et al. [33, 34] used a four-ball tribo-tester to evaluate COF and WSD of SAE-40 and mineral lubricating oils with a 30% palm–sesame biodiesel (50% palm oil, 50% sesame oil) blend with 70% diesel and 100 ppm TiO<sub>2</sub> or 20% DMC as additives and found improved lubricity with TiO<sub>2</sub> blended fuel and less COF, less WSD, and good engine and emission performance of DMC diluted 30% 50P50S–diesel fuel.

In this study, a 30:70 ratio of *Litchi-chinensis* biodiesel and diesel was used to prepare the biodiesel–diesel blend (B30). To improve the fuel characteristics, such as cold start performance, engine wear, and emissions, a novel approach of investigating safe concentrations, 10%–15% by volume of an economical oxygenated additive, EE, and 100 ppm aluminum oxide (Al<sub>2</sub>O<sub>3</sub>, abbreviated as Al in the following text) nanoparticles combined with 30% *Litchi-chinensis* biodiesel–diesel blend was used. The main objectives of this work were to develop innovative blends of litchi biodiesel with EE and Al as fuel additives and examine these blends for improvements in the engine's performance and emission characteristics and tribological properties. These tests provide practical insights into optimizing these blends for real-world applications.

## 1.1 | The Novelty of the Present Work

Biodiesel derived from various feedstocks, additives to enhance their properties, and results from engine testing are discussed in Section 1. However, studies specifically addressing *Litchi-chinensis* seed oil (LSO) biodiesel are only limited to oil extraction and properties. This research aims to cover the gap by experimentally testing *Litchi-chinensis* biodiesel produced from fast and efficient microwave-assisted transesterification with a waste-derived catalyst for promoting sustainable and clean energy in alignment with UN SDGs 7 and 13.

The combination of binary (LSO biodiesel–diesel) and ternary (LSO biodiesel–diesel–EE) blends mixed with Al nanoparticles in the specific compositions used in this study has not been explored in past works. The study is focused on determining

engine performance, combustion, and emission characteristics. In addition to this, experimental investigation also includes tribological aspects of various blends.

## 2 | Materials and Methods

### 2.1 | Biodiesel: Production and Recovery

The microwave-assisted transesterification process was used to produce biodiesel from LSO. Time, stirrer speed, methanol concentration, and catalyst concentration all impact the biofuel production from this process. The catalyst was extracted by calcination of 2 kg of dry potato waste, and the potassium hydroxide (KOH) catalyst was recovered from the ash solution. To maintain the temperature at 64°C, the microwave oven was run at 30% power. Glycerin separation occurred when the solution was stored for 8 h. After that, the biodiesel was heated to 100°C for 30 min to remove any remaining water content and later washed with hot distilled water. The properties of the LSO biodiesel product are presented in Table 2. A brief pictorial representation of the process is displayed in Figure 1. Further details on the experimental parameters and equipment details can be found in our previous work, focusing on the biodiesel production [18].

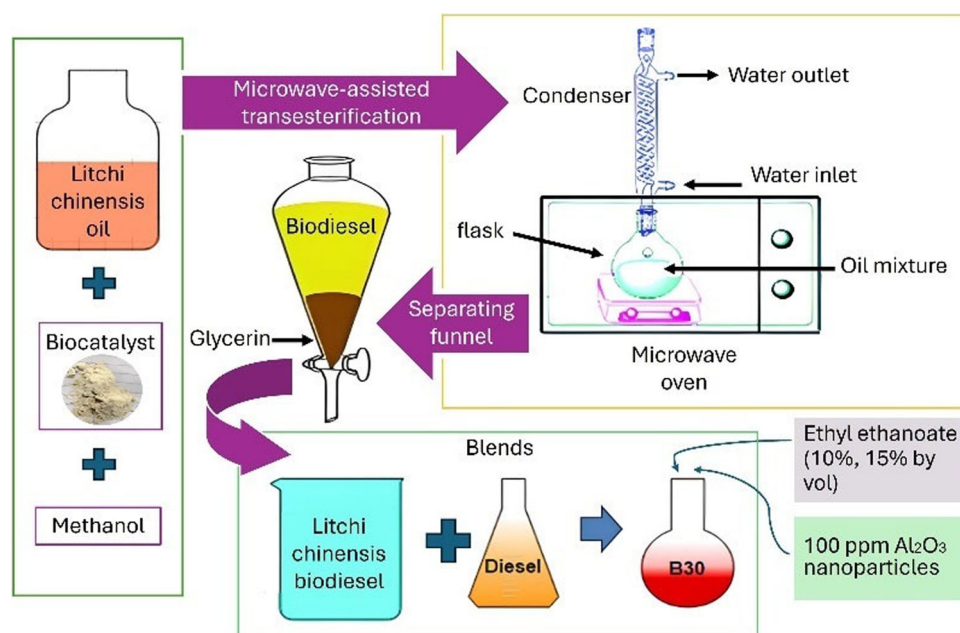
### 2.2 | Preparation of Fuel Blends

In addition to commercial diesel, six blends were prepared in this work. The fuel blend compositions are given in Table 3 along with the nomenclature to be used throughout the study. For fuel blend compositions, the basic biodiesel–diesel blend was formed in 30–70 proportions, thus called B30 throughout the following text. The primary sample size taken for the B30 blend is 1000 and 100 g for each engine and tribological testing cycle, respectively. Initially, diesel and biodiesel prepared with transesterification were mixed in a container, and the solution was stirred well. Without using heat, the mixture was agitated for 10 min. After a uniform B30 blend was formed, the alcoholic additive EE and the Al nanoparticles (20–30 nm, spherical shape) [73] were added in the decided proportions. The prepared fuel blends were then simultaneously heated and subjected to mechanical stirring using a hot plate magnetic stirrer (Heidolph MR Hei-End) for 10 min at 1000 revolutions per minute (rpm) and at a temperature of 60°C to achieve an initial uniform distribution of nanoparticles. This step ensures that larger agglomerates are broken down and dispersed throughout the blend to get homogenized ternary blends (diesel + biodiesel + EE) [26].

Following mechanical stirring, the blends were treated using an ultrasonic bath (Elma E100H), operating at 37 kHz for 60 min. Sonication [74] was done to ensure the dispersion of aluminum oxide nanoparticles in B30 and B30 + EE blends. This technique agitated the solution with sound waves, which aided in the formation of a uniform dispersion of the mixed solution. Ultrasonication is widely recognized for its ability to deagglomerate nanoparticles and achieve a stable colloidal suspension by generating cavitation forces that further break down any

**TABLE 2** | The properties of LSO biodiesel compared to international standards.

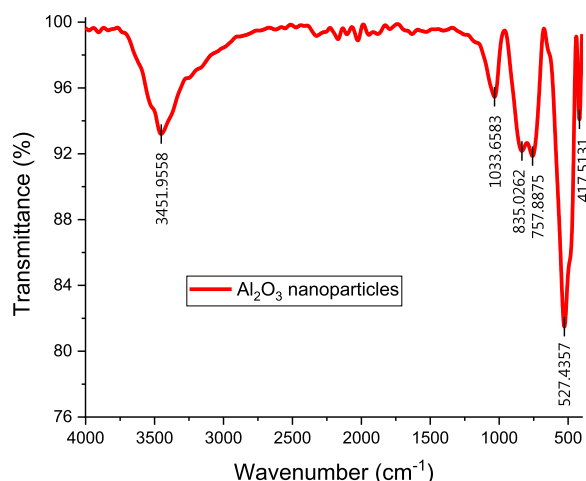
| Properties                 | Units              | WCO biodiesel [69] | Diesel [70] | LSO B100 [18] | ASTM D6751 [71] | EN14214 [72] |
|----------------------------|--------------------|--------------------|-------------|---------------|-----------------|--------------|
| Oxidative stability        | h                  | 4.61               | 13.20       | 6.1           | 3               | 6            |
| Cetane no.                 |                    | 61                 | 48.50       | 53.2          | > 47            | > 51         |
| Calorific value            | MJ/kg              | 38.76              | 46.6        | 38.8          | —               | —            |
| Acid value                 | mg KOH/g           | 0.13               | 0.15        | 0.25          | < 0.50          | 0.50         |
| Viscosity at 40°C          | mm <sup>2</sup> /s | 5.01               | 2.91        | 4.4           | 1.9–6           | 3.5–5        |
| Density at 15°C            | kg/m <sup>3</sup>  | 862.1              | 842         | 875           | 870–900         | 860–900      |
| Flashpoint                 | °C                 | 154                | 78.50       | 175           | > 130           | > 120        |
| Pour point                 | °C                 | 12                 | 2           | 0.3           | –15 to 16       | —            |
| Cloud point                | °C                 | 2                  | 2           | 6.59          | –3 to 12        | —            |
| Cold filter plugging point | °C                 | 7                  | 0           | –2.34         | 19 (max)        | —            |

**FIGURE 1** | Schematic display of biodiesel development and blend formation process. The litchi seed oil presented on the left went through esterification to transform into biodiesel. That was then mixed with diesel to form B30 and further blends (concentrations are mentioned in Table 3).**TABLE 3** | Fuel blend compositions of diesel, *Litchi chinensis* biodiesel, oxygenated additive EE, and Al nanoparticles.

| Fuel blends | Biodiesel–diesel % composition | EE/percentage volume to B30 blend | Al/concentration parts per million (ppm) | Nomenclature |
|-------------|--------------------------------|-----------------------------------|------------------------------------------|--------------|
| 1           | 0–100                          | 0                                 | 0                                        | B0 (diesel)  |
| 2           | 30–70                          | 0                                 | 0                                        | B30          |
| 3           | 30–70                          | 10                                | 0                                        | B30EE10      |
| 4           | 30–70                          | 15                                | 0                                        | B30EE15      |
| 5           | 30–70                          | 0                                 | 100                                      | B30Al        |
| 6           | 30–70                          | 10                                | 100                                      | B30EE10Al    |
| 7           | 30–70                          | 15                                | 100                                      | B30EE15Al    |

**TABLE 4** | Properties of fuel blends.

| Properties        | B0 (diesel) | B30    | B30EE10 | B30EE15 | B30Al  | B30EE10Al | B30EE15Al |
|-------------------|-------------|--------|---------|---------|--------|-----------|-----------|
| Density at 15°C   | 842         | 851    | 848.2   | 850.5   | 854    | 855       | 856       |
| Viscosity at 40°C | 2.91        | 3.471  | 2.451   | 2.105   | 3.601  | 2.721     | 2.331     |
| Calorific value   | 46.6        | 40.671 | 39.162  | 38.352  | 38.562 | 38.236    | 37.841    |

**FIGURE 2** | FTIR spectrum of Al nanoparticles.

remaining clusters of Al nanoparticles. Al nanoparticles are used in 100 ppm, while EE is added in 10% and 15% by volume. The blends were investigated for the dispersion stability of the Al nanoparticles, and no significant agglomeration or sedimentation was noticed over 7 days. This ensures that the nanoparticles are highly stable within the host fluid. This increased stability of the nanoparticles accounts for the series of schemes followed, that is, magnetic stirrings followed by ultrasonication, which broke down the nanoparticle aggregates within the host fluid and made their uniform dispersion possible. Also, the nanoparticles are spherical in shape and more stable in high-viscosity fluids compared to low-viscosity fluids [75]. This is another parameter that ensures their long-term stability. The presence of EE in the fuel blend may further enhance dispersion stability due to its polarity and solvent properties, which help reduce nanoparticle aggregation by modifying the surface tension and improving compatibility between the fuel components and nanoparticles [76]. Given that the prepared fuel blends were used within 48 h of preparation for all engine and tribological testing, the observed stability during this time frame ensured reliable and repeatable results.

Before engine test performance, a few properties including CV, density, and viscosity, were determined for fuel blend samples to observe the effects of additives. These properties are exhibited in Table 4.

### 2.3 | Characterization of Nanoparticles

Al nanoparticles used in this study are sourced from Nanos-structured & Amorphous Materials Inc., USA, and have an average particle size of 20–30 nm and spherical morphology. The FTIR

**TABLE 5** | Engine specifications.

| Factors                           | Values            |
|-----------------------------------|-------------------|
| Cylinders                         | 03                |
| Stroke                            | 04                |
| Compression ratio                 | 16.5              |
| Displacement                      | 2.5 L             |
| Max. power                        | 50 Hp @2250 rpm   |
| Max. torque                       | 172 N m @1400 rpm |
| Static injection timing           | 16 degree BTDC    |
| Nozzle opening/injection pressure | 190 bar           |
| Lubricant capacity                | 6/17 L            |
| Cooling                           | Water type        |

spectrum of sample nanoparticles is shown in Figure 2, which gives significant information about the bond structure and composition of nanoparticles under study. The spectroscopy was performed for the 4000–400  $\text{cm}^{-1}$  wavelength range. The noteworthy infrared transmittance peaks can be observed at 3451.96, 1033.66, 832.133, 757.888, 527.436, and 417.513  $\text{cm}^{-1}$ . The first peak at 1033.66  $\text{cm}^{-1}$  with a transmittance of 95.46% indicates the presence of a hydroxyl group at the surface due to moisture. A characteristic peak at 1033.66  $\text{cm}^{-1}$  confirms it to be an oxide. At 832.133 and 757.888  $\text{cm}^{-1}$ , two fingerprint peaks are attributed to metal-oxide bond vibration in  $\text{Al}_2\text{O}_3$ . The sharp peak at 527.436  $\text{cm}^{-1}$  and distinctive peak at 417.513  $\text{cm}^{-1}$  are due to lattice vibrations of the Al–O–Al bond, which is a signature characteristic of  $\text{Al}_2\text{O}_3$ .

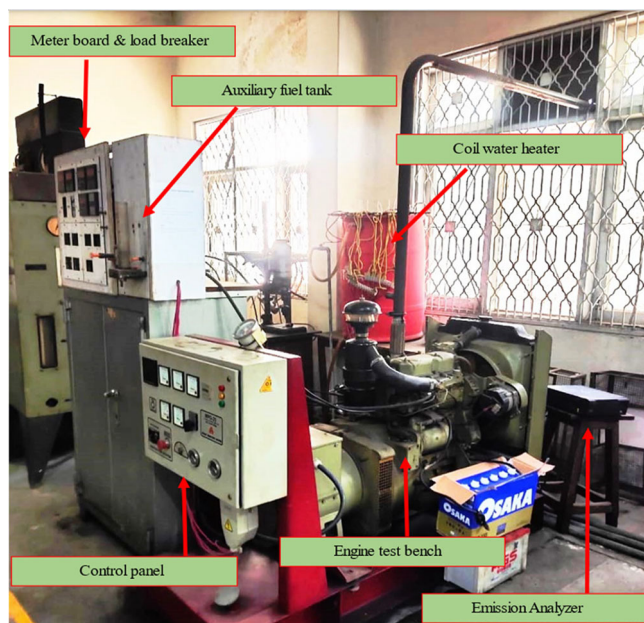
### 2.4 | Experimental Setup: Engine and Emission Analyzer

The experimental setup consists of a diesel engine (Perkins/AD 3.152) test bench loaded with a series of water heaters. The auxiliaries include a fuel tank, control panel, startup battery, an emission monitoring system (EMS) analyzer, and a display board with various load breakers. The water heaters act as a load on the system by receiving its voltage and current from the engine on the test bench. Specifications of the engine used for the experimentation are displayed in Table 5.

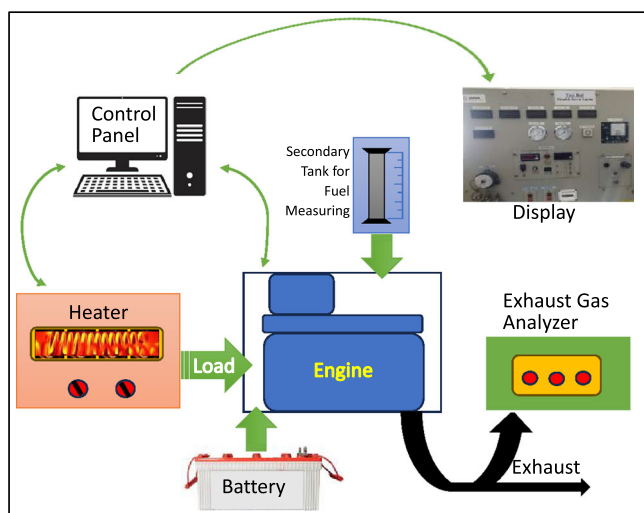
Figure 3 shows a model picture, and Figure 4 is a schematic representation of the test engine configuration. Engine testing was done by increasing engine speed at intervals of 100 rpm from 1000 to 1500 rpm while maintaining a constant load of 4500 W. To keep the engine running on the required load

conditions, a total of nine heaters were powered from the test engine setup, each consuming 500 W of energy.

An auxiliary fuel tank was used to get the rate of fuel consumption for each test sample of 100 mL. At the same time, the



**FIGURE 3** | The basic arrangement and components of the diesel engine test bench.



**FIGURE 4** | A schematic representation of all the main and auxiliary parts of the engine setup (dimensions not to scale).

voltage and current data from the display panel were recorded. The performance was then assessed at each of the engine speeds for the fuel blend combinations as indicated in Table 3. BSFC, BTE, and BP were calculated for performance assessment. At each engine speed, the exhaust values of HC, CO, and NO<sub>x</sub> were measured from the EMS analyzer. The specifications of the EMS are mentioned in Table 6.

## 2.5 | Four-Ball Tribo-Testing

The COF in engine functions is affected by a number of variables, including lubricant type, environmental temperature, load, speed, and additives that may enter the lube oil during engine operation. By tracking the COF over time and observing its variation during the experimentations, one may gain insight into the lubricant's performance in dynamic situations. This information is useful for determining the behavior of lubricant across a range of time intervals and may also be used to evaluate how well it reduces wear and friction over longer working times. To get that information, a four-ball tribo-tester was used in the current study. The COF was determined at various test phases by tracking the frictional torque over time, and data on the COF throughout those time periods were obtained from it. The frictional forces between the surfaces of the revolving steel ball and the stationary balls coated with lubricant, as presented schematically in Figure 5, were monitored intermittently during the test.

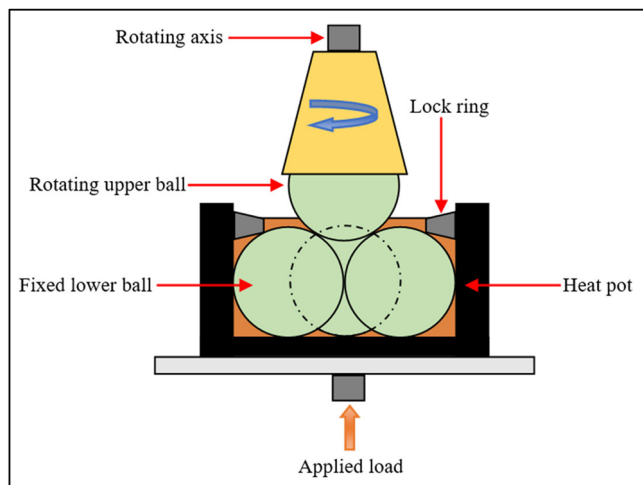
The effect of 5% of all seven fuel blends on the COF of SAE-40 base lubricant was investigated using an automatic four-ball tribo-tester (MMW-1, Computer Servo Control Vertical Universal Friction and Wear Testing Machine). Four carbon-chromium steel balls were submerged in the mix of 95% SAE-40 standard lubricant oil and 5% sample fuel blends. Three fixed steel balls were attached to a temperature sensor and placed in a cup holder. The fourth ball was adjusted on the top rotating shaft. The temperature of the setup was kept constant when the speed of the rotating shaft was 1200 rpm. A new set of four carbon-chromium steel balls was used for each repeated cycle of the experimentation as all seven of the fuel-lubricant mixtures were tested under the ASTM D4172 standard. The four-ball tribo-tester's operating parameters under given conditions are presented in Table 7.

## 2.6 | Calculations: BSFC, BTE, and COF

Fuel blends were tested for the quantity of 100 mL samples for each measurement. The rate of fuel consumed ( $\dot{m}$ ) was noted

**TABLE 6** | Specifications of EMS gas analyzer.

|              | Technique used                                      | Emission type   | Measurement range | Resolution    |
|--------------|-----------------------------------------------------|-----------------|-------------------|---------------|
| EMS analyzer | NDIR analyzer (nondispersive infrared)              | CO              | 0%–10% vol.       | ± 0.001% vol. |
|              | CLD-type (chemiluminescence detector) heated vacuum | NO <sub>x</sub> | 0–5000 ppm        | ± 1 ppm       |
|              | FID (flame ionization detector)                     | HC              | 0–9999 ppm        | ± 1 ppm       |



**FIGURE 5** | A schematic of a four-ball tribo-tester.

from the auxiliary fuel tank scale for each running cycle. BP is calculated by Equation (1), using voltage and ampere values from the display panel and the manufacturer's provided values of power factor ( $\phi$ ) and generator efficiency ( $\eta_g$ ) for equipment.

$$BP = \frac{(V_1 I_1 + V_2 I_2 + V_3 I_3) \cos \phi}{\eta_g}, \quad (1)$$

whereas  $V_1$ ,  $V_2$ ,  $V_3$  and  $I_1$ ,  $I_2$ ,  $I_3$  represent the voltage and current readings, respectively. Subscripts of voltage and ampere notations indicate the repetitive testing to minimize the error.

$$BSFC = \frac{\dot{m}}{BP}, \quad (2)$$

$$BTE = \frac{BP \times 3600}{\dot{m} \times CV}. \quad (3)$$

The generator efficiency varies from 80% to 85%, depending upon the electric power generated, and the power factor value is 0.85 for the experimental setup. Equations (2) and (3) were used for further calculations to obtain the data sets of engine performance evaluation for each fuel blend of known CV.

### 3 | Results and Discussions

#### 3.1 | Engine Performance

Figure 6a shows the experimental results of BSFC for the diesel and six tested blends over a range of engine speeds. The overall results reveal that BSFC is higher at lower engine speeds, suggesting that engine operating conditions play a direct role in determining BSFC. Figure 6a shows that at 1100, 1200, 1300, and 1500 rpm, the B30EE10 blend and, at 1400 rpm, the B30Al blend exhibit the lowest BSFC values. It is interesting to note that at 1000 rpm, the BSFC of both blends is comparable to that of the B0 (100% diesel). Compared to diesel, B30EE15Al, B30EE10Al, and B30 show little to no improvement in terms of BSFC. When averaged over the engine speeds (see Figure 6b), the B30EE10 and B30Al blends show a 6.21% and 5.54%

**TABLE 7** | ASTM D4172 testing standard for four-ball tribo-tester.

| <i>Ball's specification</i>        |                       |
|------------------------------------|-----------------------|
| Material of the tested balls       | Carbon–chromium steel |
| Hardness of the tested balls       | 62                    |
| Diameter/size of the tested balls  | 12.7 mm               |
| Roughness of ball's surface        | 0.04 mm               |
| <i>Operating factors</i>           |                       |
| Upward load under the heat pot     | 40 kg                 |
| Spindle rotational speed           | 1200 rpm              |
| Constant temperature maintained at | 75 °C                 |
| Require time for each test sample  | 60 min                |

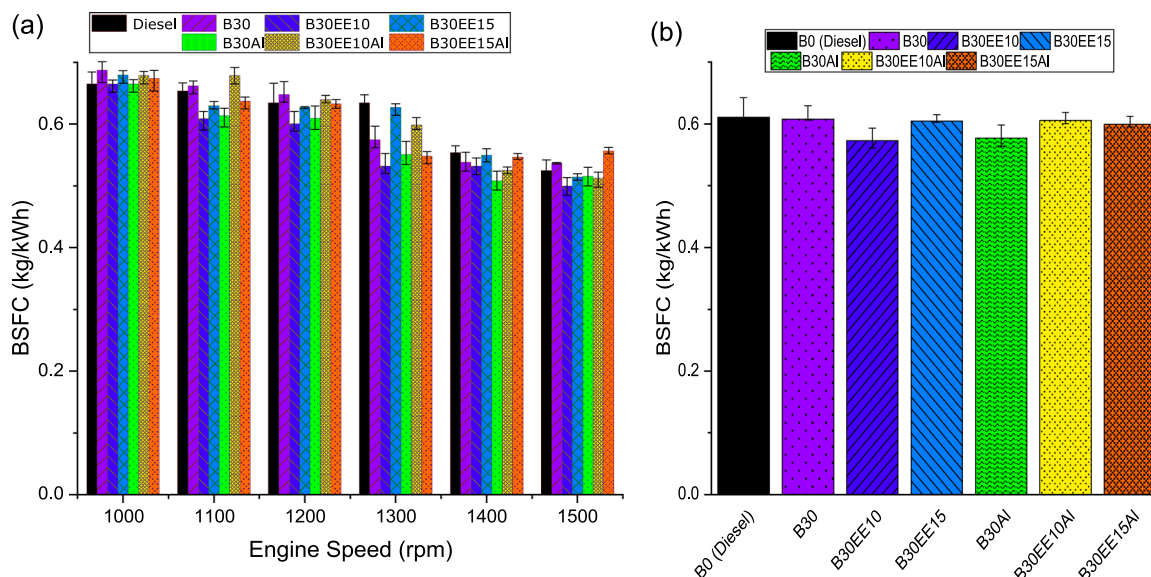
decrease over B0, respectively. Other fuel combinations also exhibit modest improvements in BSFC, highlighting the significance of optimizing the B30 blends with oxygenated additives (EE) and Al nanoparticles to attain the best fuel consumption efficiency. These results are consistent with the previous studies where BSFC has an overall decreasing trend with engine load and speed [28, 32, 33].

Figure 7 shows BTE for the diesel and tested blends over the tested engine speed range. In Figure 7a, the overall trend shows an increase in BTE with the increase in engine speed. Figure 7b shows that, when averaged over the engine speed, all blends show higher BTE when compared to diesel, except B30. A decrease of 4.86% in BTE is observed for B30. Other than the B30 blend, the inclusion of Al and EE has significantly improved the engine performance with increasing engine speed. B30EE10 has a higher increase in BTE, 6.02% compared to only 4.1% in B30EE15 which has a higher quantity of oxygenated additive. BTE is increased by a maximum of 6.66% for B30Al fuel blend. For EE and Al nanoparticles blend, the increase in BTE is 4.55% and 3.88% for B30EE10Al and B30EE15Al, respectively. This is consistent with previous studies given in [28, 32].

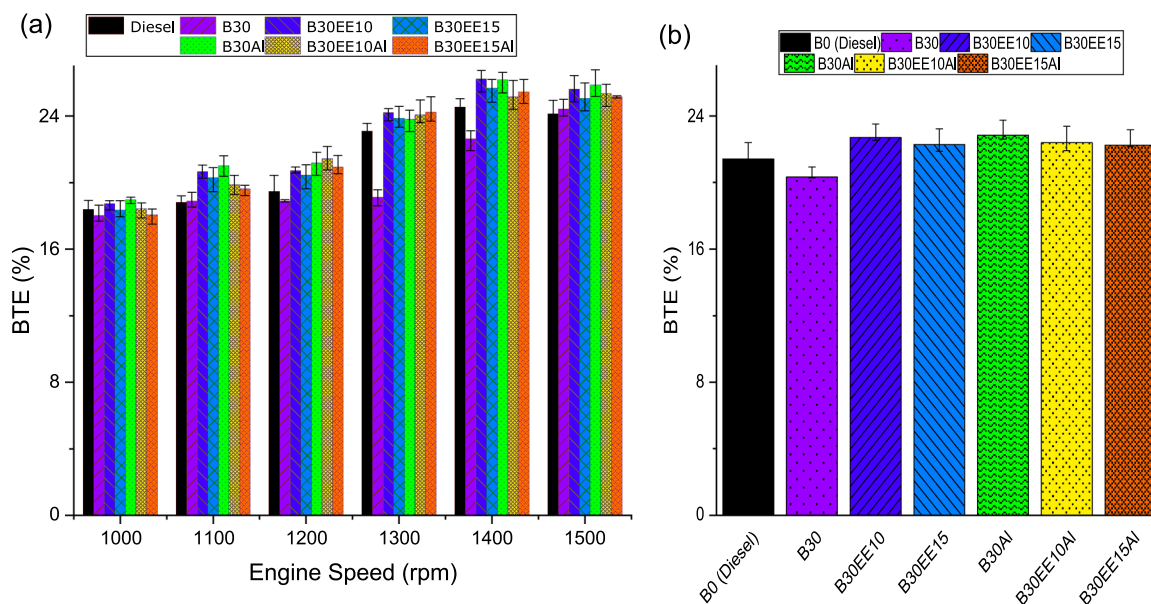
These results and their comparison with past work emphasize the need for further detailed systematic investigation to determine the best blending ratios for improving engine efficiency and reducing fuel consumption.

#### 3.2 | Engine Emissions

Emissions are generally expected to decrease in compression ignition engines using biofuels, nanoparticles, and alcoholic additives. However, poor evaporation and atomization can lead to more CO and HC emissions as a result of improper combustion [77]. Figure 8a shows the CO concentration in the exhaust gases for all tested fuels at tested engine speeds. Figure 8b shows the CO concentrations averaged over the engine speeds.



**FIGURE 6** | Engine testing results of (a) brake-specific fuel consumption (BSFC) for six different speeds and (b) mean BSFC for each fuel averaged over all speeds.



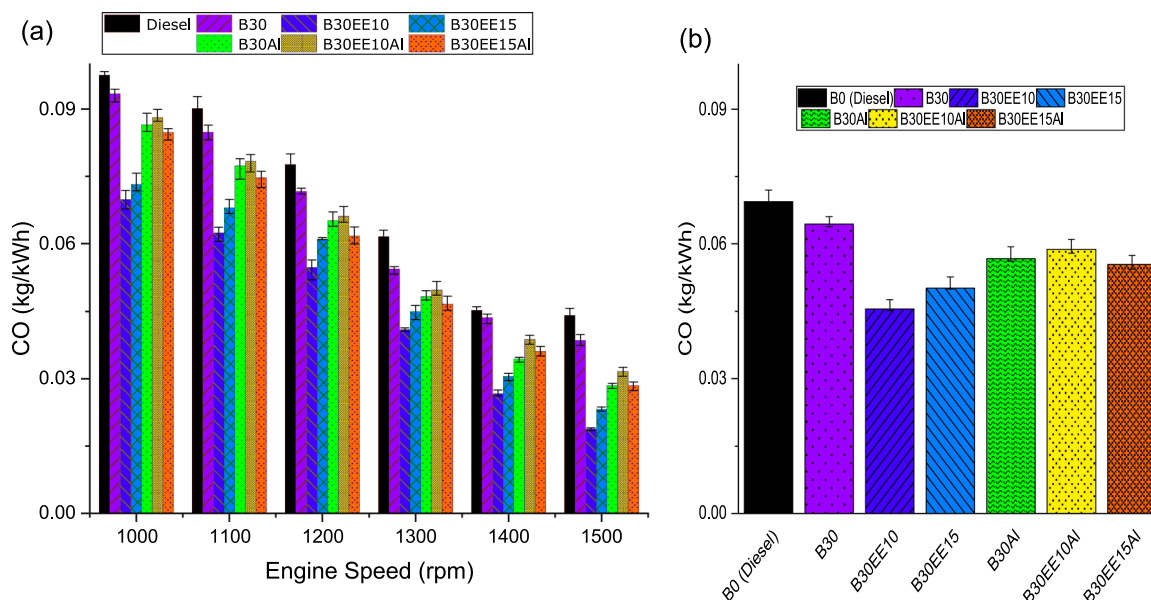
**FIGURE 7** | (a) Brake thermal efficiency (BTE) of engine tests performed at various speeds and (b) mean BTE for seven different samples of fuel blends.

CO is lower at higher speeds for all blends, which is consistent with the study by Mujtaba et al. [33].

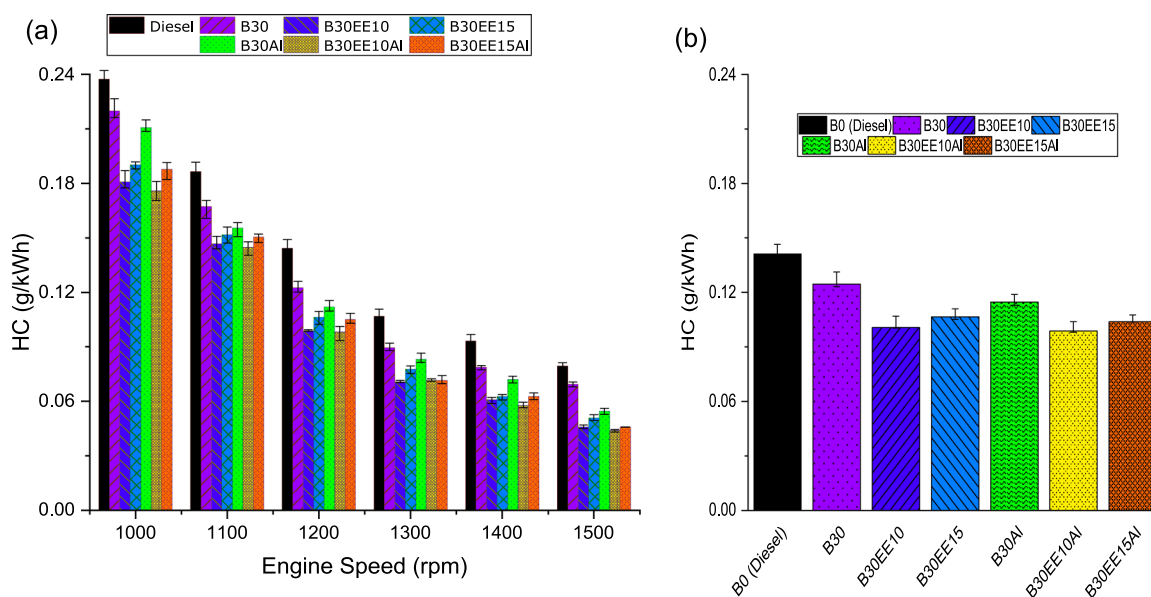
All the investigated fuel blends have shown better CO emissions than diesel. B30 and B30Al blends improve the exhaust quality, but the combination of EE and Al in blends deviates from the expected behavior. CO emissions decreased by 7.17% for B30. The blends with Al nanoparticles have shown a decrease of 18.27%, 15.27%, and 20.14% with increasing EE concentrations of 0%, 10%, and 15%, respectively. Overall, EE inclusion improves CO emissions most significantly, with B30EE10 and B30EE15 showing a reduction of 34.49% and 27.69%, respectively. This indicates that EE inclusion is effective in reducing CO emissions but up to a specific concentration. This trend is

different from what was observed with Al nanoparticles, where 10% EE showed a lower decrease in CO concentration. Overall, CO emissions are least with alcoholic additives and relatively less with nanoparticles and B30 compared to diesel.

Figure 9 shows the HC emissions for increasing speed that are similar to CO emissions in Figure 8. Across the range of tested engine speeds, HC emissions show a stable downward gradient, with diesel having the highest HC production and B30 following closely after. The reduction of HC emissions in increasing order is by using B30Al, B30EE15, B30EE15Al, B30EE10Al, and B30EE10. B30EE15 and B30EE15Al showed a reduction of 26.44% and 24.55%, respectively. Interestingly, blends with 10% EE produce the least HC in exhaust gas at all engine speeds.



**FIGURE 8** | CO emission concentration for all tested fuel samples (a) at various engine speeds. (b) Mean of CO for each fuel blend.



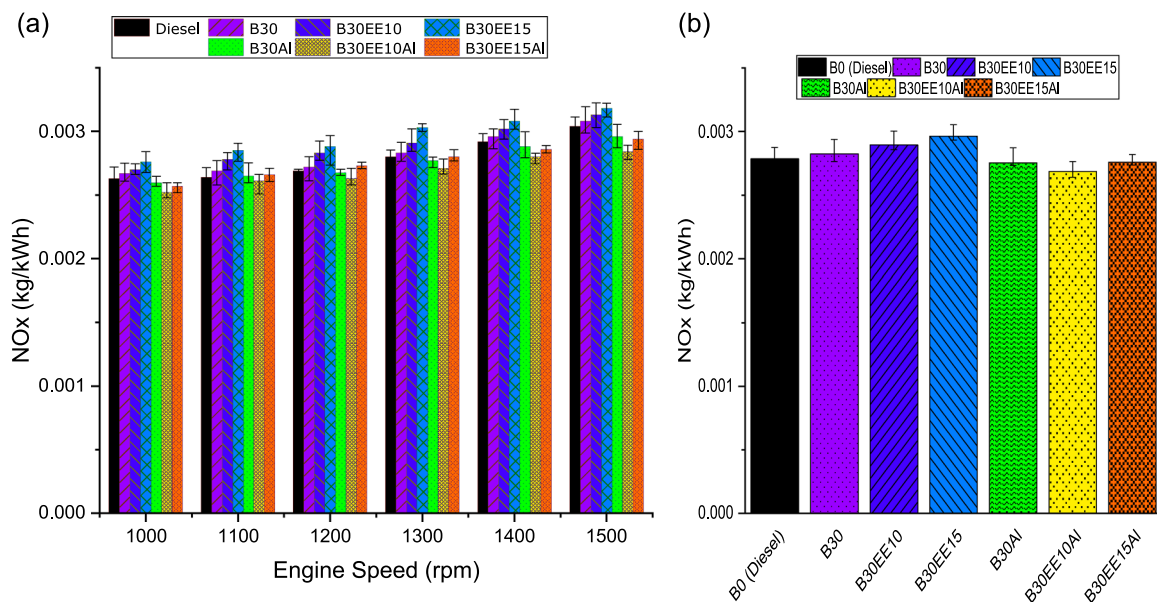
**FIGURE 9** | HC emissions for various fuel samples (a) for six different speeds and (b) the mean of HC emissions.

The B30EE10Al blend exhibits the largest reduction in HC emissions, with a 30.24% decrease. Associating these results with literature gives similar trends with speed or load conditions. Commonly, HC in exhaust amounts to the lowest with the addition of oxygenated or alcoholic additives [33]; 100 ppm Al nanoparticle inclusion shows a prominent decrease in HC emissions compared to neat diesel as well as blends having lower Al nanoparticle concentrations [32].

Smog has been an alarming concern in Southeast Asia in the past few years.  $\text{NO}_x$  formation is a contributing factor to smog [7]. The use of novel biodiesels has contributed to  $\text{NO}_x$  reduction in some cases while not in others. A study exploring four novel biodiesels revealed that more or less all four alternative fuels have contributed to lowering  $\text{NO}_x$  compared to neat diesel [7], whereas research by the same author involving waste plastic

biodiesel and Spirulina microalgae biodiesel reveals opposite results [78].

Figure 10 shows  $\text{NO}_x$  emission characteristics of the investigated fuels. Generally, the measure of  $\text{NO}_x$  emission intensity expands with increasing loads [46]. We can observe a minute general upward trend in the  $\text{NO}_x$  emissions with engine speed at a constant load of 4500 W. When averaged over the engine speeds (see Figure 10b), the blends with Al nanoparticles exhibit lower  $\text{NO}_x$  emissions compared to diesel. The lowest  $\text{NO}_x$  emission can be seen in the B30EE10Al blend, which is 3.65% less than neat diesel. B30Al and B30EE15Al blends also show a slight decrease of 1.14% and 0.95% in  $\text{NO}_x$  production, respectively. B30 showed an increase of 1.07% in  $\text{NO}_x$  emissions which increased further with EE inclusion to 3.82% and 5.98% for B30EE10 and B30EE15 respectively. In comparison with other



**FIGURE 10** | Results for NO<sub>x</sub> emission in combustion of litchi biodiesel and its blends (a) at various speeds and (b) mean of NO<sub>x</sub> value for tested fuel blends.

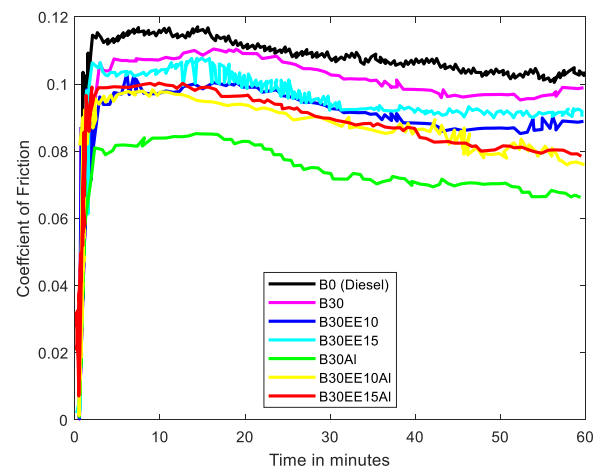
emission parameters, NO<sub>x</sub> values have not lowered much in this study.

EE has a lower CV than other fuels and hence lowers combustion temperatures, but its addition with other fuels is found to have a negative impact on NO<sub>x</sub> emissions. EE's negative impact on NO<sub>x</sub> is due to its high oxygen content. The combustion of the fuel blend improves with more oxygen, leaving a high temperature in the engine [79]. This improves the overall exhaust quality but enables the NO<sub>x</sub> formation. The oxygenated additive EE seems to lower NO<sub>x</sub> in diesel, but when studied with biofuel–diesel blends, it causes a negative effect on exhaust. This behavior is also backed by a study by Khalife et al. [39]. However, the reduction of NO<sub>x</sub> is evident to some extent when EE and aluminum oxide are combined in the B30EE10Al blend.

The experimental results highlight the intricate interactions between blend compositions and how they affect diesel engine emissions. It is necessary to explore more blend compositions to get optimized results of engine performance to achieve considerable reductions in emissions without sacrificing the fuel efficiency or the engine performance. These findings support further efforts that can be made to lessen the harms of fuel combustion in engines and create sustainable transportation fuels from biomass for green energy initiatives.

### 3.3 | Four-Ball Tribo-Tester Frictional Analysis

Figure 11 shows the COF trend for all the tested samples mixed with SAE-40 as a base lubricant over elapsed time. The COF for diesel is above 0.11 for most of the run time, which is the highest among all tested samples. Further, the lowest COF is observed for the B30Al blend, which never reached 0.09 throughout the 60 min run time. The B30 blend showed less COF compared to diesel which improves further with an

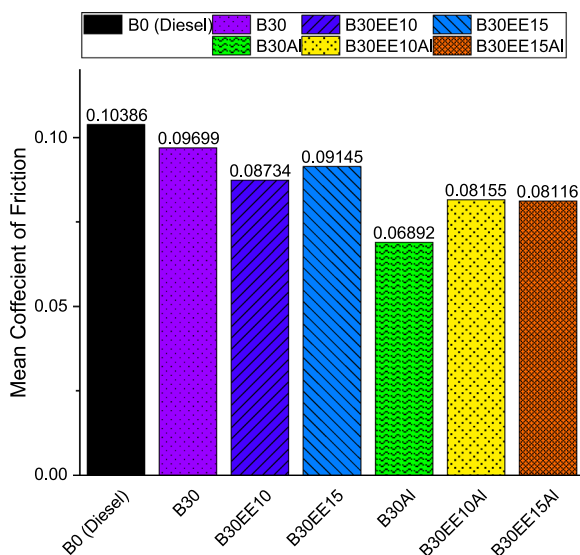


**FIGURE 11** | Variation in COF versus time.

addition of EE and Al. In this section, the fuel names represent the mixture of 5% fuel mixed with 95% SAE-40 lubricating oil.

Figure 12 shows the steady-state averaged COF for blends at the time interval of 40–60 min. Here again, B30Al has the lowest COF. From Figure 12, the B30EE10 blend is observed to have a better effect on COF reduction than the B30EE15 blend. However, with the Al addition, the average COF for the B30EE15Al blend is slightly lesser than B30EE10Al, supposedly due to sudden unstable composition or low nanoparticle miscibility.

A similar pattern is observed in the studies of palm–sesame biodiesel's COF using a four-ball tribo-tester by Mujtaba et al. [34]. It is evident that the COF was lowest for base lubricant and highest for maximum diesel and decreased with the inclusion of additives, where B30 biofuel blends have a lower COF than that of diesel [34]. COF was lower for 100% biofuel (B100) compared to B30 for most of the run time except for the initiation phase of the steady state, and average COF was lower



**FIGURE 12** | Average COF for lubricant fuel blends at steady state.

for B100 compared to B30 and its blends with additives other than metal-oxide nanoparticle additive [80]. Overall, the best results were achieved for nanoparticle-blended mixtures, and alcoholic additives have shown variations based on their individual properties. However, their average COF values are lower than B30 and diesel fuel-lubricant blends [33].

Lower COF in any biodiesel is expected due to the presence of ester molecules that adsorb on metallic surfaces, making a film between rubbing surfaces and reducing friction, and thus preventing engine wear [81]. Moreover, *Litchi chinensis* biodiesel used in the study contains high-purity methyl esters that give it more desirable frictional properties [18]. Thus, reducing the COF of the B30 blend by 6.6% on average.

The addition of EE, with its high oxygen content, seems to reduce the COF only for low concentrations. That can be due to the low kinematic viscosity of EE, which may prevent the formation of a lubricant layer on metal surfaces [47]. The adsorption of EE on vanadium carbide (VC), which is widely used as a structural component of automobile and aircraft engines and as a grain development inhibitor in hard metals, has been shown to significantly reduce frictional properties with only minor changes in interfacial adhesion. Greater loading levels and more exposure to EE increase the friction [82, 83]. This verifies the changing behavior of increased COF for high EE content blends B3015EE and B30EE15Al compared to B30EE10 and B30EE10Al in Figure 11 which shows more EE exposure leads to more adsorption and increased COF. Figure 12, on the other hand, shows a slight decrease in COF of B30EE15Al compared to B30EE10Al.

Nanoparticles have shown decreased COF regardless of EE content, as shown in Figures 11 and 12. For a complete 60-min run time, the average COF for B30EE10Al and B30EE15Al are 6% and 9% less than B30EE10 and B30EE15 respectively. The nanoparticles-biodiesel blend B30Al shows the lowest COF among all the tested samples. Al nanoparticles act as surfactants for the fuel blends and lube oils, providing better homogeneity in all kinds of mixtures [84]. Spherical Al nanoparticles convert

sliding friction between surfaces to rolling friction due to the ball-bearing effect and improved lubricating qualities by self-laminating the friction surface. As a result, in an experimental study of a four-ball test, 17.61% friction and 41.75% wear reduction were discovered for the best of 0.1% by weight of Al when testing up to 1% by weight of Al nanoparticles added to the lubricant [84]. This shows that the Al nanoparticles exhibit overall good frictional properties but only up to certain concentrations, and 100 ppm works appropriately for the 30% litchi seed biodiesel-diesel blends.

Table 8 summarizes the comparison of the present study with the findings of previous researchers' works in terms of engine performance, tailpipe emissions, and tribological parameters.

## 4 | Conclusion

Using a microwave-assisted transesterification method, *Litchi chinensis* biodiesel was created from LSO. The combination of 30% biodiesel and 70% diesel (B30) was mixed with EE in 10% (B30EE10) and 15% (B30EE15) by volume for better oxygenation of litchi biodiesel. For further improvement of these blends, 100 ppm  $\text{Al}_2\text{O}_3$  (abbreviated as Al) was added to create nanoparticle-infused blends (B30Al, B30EE10Al, and B30EE15Al). Sonication was performed to make homogenized nanoadditive blends for smooth engine performance and emission testing, whereas a four-ball tribo-tester was used for frictional analysis of fuel mixes. The conclusion is drawn based on the results that were obtained during the process.

- The B30EE10 blend reduced BSFC by up to 6.21%, while the B30Al blend reduced it by 5.54% while improving engine's BTE by 6.66% compared to B0.
- In terms of emissions, the B30EE10 blend achieved a 34.49% decrease in CO and a 28.72% reduction in HC emissions. The B30EE10Al blend gives optimal emissions, with an average 3.92% reduction in  $\text{NO}_x$  compared to B0.
- A four-ball tribo-tester was utilized to measure the COF of 95% SAE-40 lubricant and 5% of the fuel blends under the ASTM D4172 standard. The B30Al fuel blend has shown the lowest mean COF value of 0.068, which is 33% lower than the mean COF for B0.
- $\text{Al}_2\text{O}_3$  induced better frictional properties compared to oxygenated additives in the litchi seed biodiesel-diesel contaminated lube oil. However, blends with EE express intense COF variation, indicating blend instability compared to SAE-40 lubricant.

Overall, adding 100 ppm Al nanoparticles and 10% EE to 30% biodiesel (B30) made from LSO significantly improved the fuel's quality. These improvements were especially noticeable in terms of better cold flow properties (how the fuel behaves at low temperatures) and reduced emissions, making the fuel blends more efficient and environmentally friendly.

The fuel mixes containing litchi seed biodiesel, specifically the B30EE10Al blend, are recommended to be used in industries, power generation, and the transport sector as part of a pilot

**TABLE 8** | Comparison of present work with other published work.

| Author                   | Feedstock used/<br>method                                                                                           | Additives/<br>blends                                                                                                               | Engine used                                                                     | Results                                                    | Novelty                                                                                                                                                                                                                                                                                                                                                     |
|--------------------------|---------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------|---------------------------------------------------------------------------------|------------------------------------------------------------|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Mujtaba<br>et al. [27]   | 50–50 palm–sesame<br>oil/ultrasound-<br>assisted<br>transesterification                                             | B30<br>B30 + DMC<br>B30 + DEE<br>B30 + CNT<br>B30 + TiO <sub>2</sub>                                                               | 1-Cylinder,<br>0.638 L, CR:<br>17.7, radiator-<br>cooled diesel<br>engine       | BT↑<br>BTE↑<br>HC↓<br>CO↓<br>NO <sub>x</sub> ↑             | A mixture of two oils<br>(palm–sesame) with harmonizing<br>properties is utilized for biodiesel<br>production and studied for the<br>effects of alcohols and<br>nanoparticle additives                                                                                                                                                                      |
| Pullagura<br>et al. [85] | Sea mango oil/<br>catalytic<br>transesterification                                                                  | SMME15<br>TF<br>TF + SiO <sub>2</sub> 30<br>TF + SiO <sub>2</sub> 60<br>TF + SiO <sub>2</sub> 90                                   | 1-Cylinder,<br>direct<br>injection,<br>liquid-cooled<br>diesel engine           | BSFC↓<br>BTE↑<br>CO↓<br>HC↓<br>NO <sub>x</sub> ↓<br>Smoke↓ | The study investigates the effect<br>of adding metal-based SiO <sub>2</sub><br>nanoparticles in concentrations<br>of 30, 60, and 90 mg/L in ternary<br>fuel (TF) mix<br>*TF = 75% diesel + 15% sea<br>mango methyl ester<br>(SMME15) + 10% <i>iso</i> -butanol                                                                                              |
| Devarajan<br>et al. [38] | Papaya seed oil/<br>transesterification in a<br>5-L batch reactor with<br>NaOH as catalyst                          | Diesel<br>BD50<br>BD50 + n-D<br>BD50 + DTBP                                                                                        | A four-stroke,<br>single-<br>cylinder diesel<br>engine                          | BSFC↑<br>BTE↓<br>NO <sub>x</sub> ↑<br>HC↓<br>CO↓<br>Smoke↓ | Two oxygenated additives in a<br>biodiesel–diesel blend are<br>studied in diesel engine at<br>variable brake mean effective<br>pressure (1–5 bar) and a constant<br>speed of 1800 rpm<br>*n-D = decanol,<br>DTBP = ditetrabutylperoxide                                                                                                                     |
| Aalam<br>et al. [58]     | Ziziphus jujube oil/<br>catalytic<br>transesterification                                                            | Diesel<br>ZJME (25%–75%<br>biodiesel–diesel)<br>ZJME + 25Al <sub>2</sub> O <sub>3</sub><br>ZJME + 50Al <sub>2</sub> O <sub>3</sub> | A single<br>cylinder,<br>common-rail<br>direct<br>injection<br>(CRDI)<br>system | BTE↑<br>Smoke↑<br>SFC↑<br>NO <sub>x</sub> ↑<br>CO↓<br>HC↓  | This study investigates the impact<br>of a novel biodiesel, 25% blended<br>with diesel, enhanced with 25 and<br>50 ppm aluminum oxide<br>nanoparticles on combustion,<br>performance, and emissions of a<br>CRDI diesel engine                                                                                                                              |
| Present<br>work          | <i>Litchi chinensis</i> seed<br>oil/microwave-<br>assisted<br>transesterification<br>using a bio-<br>catalyst (KOH) | B0 (diesel)<br>B30<br>B30EE10<br>B30EE15<br>B30Al<br>B30EE10Al<br>B30EE15Al                                                        | A four-stroke,<br>three-cylinder<br>diesel engine                               | BSFC↓↑<br>BTE↑<br>HC↓<br>CO↓<br>NO <sub>x</sub> ↓↑<br>COF↓ | Novel use of <i>Litchi chinensis</i><br>biodiesel mixed with oxygenated<br>additive ethyl ethanoate<br>(10%–15%) and Al <sub>2</sub> O <sub>3</sub><br>nanoadditives sonicated in both<br>biodiesel–diesel and ternary<br>blends. The study focused on<br>engine performance, emissions, as<br>well as the tribological benefits of<br>various fuel samples |

project aimed at gauging the blend qualities for long-term use in engines, specifically in sectors where engine usage is inevitable. However, only with effective waste management and controlled processing of the litchiplant's seed waste can economical and green production be made possible.

The integration of diesel with biodiesel and additives for testing new biofuels promotes further research on sustainability by utilizing agricultural waste, reducing emissions, and advancing organic waste management in the transportation sector. Technological developments in nonfood biomass biodiesels have the potential to increase their sustainability and lessen competition with food supply. All things considered, biofuel research is

essential to advancing clean energy, safeguarding a stable and varied energy future, and preserving the environment.

#### Nomenclature

| Symbol                         | Description                                          |
|--------------------------------|------------------------------------------------------|
| EE                             | ethyl ethanoate/ethyl acetate                        |
| Al <sub>2</sub> O <sub>3</sub> | aluminum oxide                                       |
| Al                             | 100 ppm aluminum oxide nanoparticles                 |
| B30                            | 70% diesel and 30% <i>Litchi chinensis</i> biodiesel |
| EE10                           | 10% ethyl ethanoate by volume of a blend             |

|                 |                                          |
|-----------------|------------------------------------------|
| EE15            | 15% ethyl ethanoate by volume of a blend |
| LSO             | litchi seed oil                          |
| WCO             | waste cooking oil                        |
| KOH             | potassium hydroxide catalyst             |
| BSFC            | brake-specific fuel consumption          |
| SFC             | specific fuel consumption                |
| BP              | brake power                              |
| BTE             | brake thermal efficiency                 |
| CO              | carbon monoxide                          |
| HCS             | hydrocarbons                             |
| NO <sub>x</sub> | oxides of nitrogen                       |

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